





Development of new pretreatment approach for food safety screening

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Food safety has been a global public health concern. Humans are seriously endangered by numerous contaminants in foodstuffs. Owning to the complexity of food matrices and trace level of food

contaminant, sample pretreatment is an essential step in food safety screening (sensitivity, accuracy and analytical speed)

The available approaches (e.g. solid phase extraction (SPE), solid phase microextraction (SPME))often show some limitations such as time consuming, complicated operation, low affinity and selectivity to targeted analytes.



- The rapid development of the nanotechnology has brought many opportunities for food sample pretreatment.
- Nanomaterials were synthesized from organic or inorganic materials, and their typical size range is between about 0.2-100 nm Nanomaterials are considered as great adsorbents due to their ultrasmall size, large surface area, unique structure, and functional properties, allowing for the efficient isolation and pre-concentration of contaminants from foodstuffs



The application of nanomaterials as adsorbents has become a promising trend in the field of food safety screening. Diverse types of nanomaterials have been evaluated in food sample pretreatment, such as metal-organic frameworks (MOFs), ordered mesoporous silicas (OMSs), polydopamine-derived materials (PDA), carbon-based materials, molecularly imprinted polymers (MIPs) as well as other novel nanomaterials.

The emerging nanomaterials have presented a better performance for the extraction and pre-concentration of food contaminants, which significantly improve the detection sensitivity and selectivity.



Covalent organic frameworks (COFs) are an emerging group of microporous materials. COFs have attracted increasing attention in sample pretreatment due to their properties such as high surface area, tunable pore size, good chemical selectivity and thermal stability.

Recently, we developed a serious of magnetic COFs for enrichment of Polycyclic Aromatic Hydrocarbons (PAHs), plant growth regulators (PGRs) and fluoroquinolones (FQs)



1. Effective Enrichment and Detection of Trace Polycyclic Aromatic Hydrocarbons in Food Samples based on Magnetic Covalent Organic Framework Hybrid Microspheres



Schiff base reaction Hydrothermal synthesis process



TEM



Topology





Effective Enrichment and Detection of Trace Polycyclic Aromatic Hydrocarbons in Food Samples based on Magnetic Covalent Organic Framework Hybrid Microspheres





Optimization of conditions

Amount of COFs: 5 mg; elution : Acetonitrile; enrichment time: 15 min; elution time: 10 mins



Typical chromatograms obtained after MSPE from spiked real samples. Peak identification: **1**, naphthalene; **2**, acenaphthylene; **3**, fluorene; **4**, phenanthrene; **5**, anthracene; **6**, fluoranthene; **7**, pyrene; **8**, perylene; **9**, benzo[*a*]anthracene; **10**, chrysene; **11**, benzo[*b*]fluoranthene; **12**, benzo[*k*]fluorathene; **13**, benzoapyrene; **14**, dibenz[*a*,*h*]anthracene; **15**, benzo[*g*,*h*,*i*]peryrene.

HPLC-DAD ; Hypersil GOLD column (150×4.6 mm, 3µm, Thermo, USA) ; Detection limits : 0.83-11.7 ng/L ; Recovery : 86.7-104.5%

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ESI Highly cited paper

2. Sulphonate functionalized covalent organic framework-based magnetic sorbent for effective solid phase extraction and determination of fluoroquinolones (FQs)





Characterization of materials(TEM, FT-IR spectra, BET surface area)



Adsorption kinetics

Adsorption kinetics

Analytes @	$Q_{e}(mg/g)$	Pseud	o-first-order n	nodel 🛛	Pseudo-second-order model @				
		<u>K1</u> * ³	Qe,cal	R •3	$K_2 \text{e}^2$	Qe,cal	R +3		
			(mg/g) ~			(mg/g) ~			
ENO 43	8.52 +	0.029	4.02	0.9376 @	0.014	8.89	0.9912 @		
OFL +2	9.28	0.023 @	1.59~	0.8861 @	0.039*	9.37*	0.9982		
CIP 🖓	7.34 @	0.033 @	9.29	0.9318 @	0.017 ~	7.63 🛛	0.9939.		
DAN 🖉	7.24 @	0.053 @	17.40	0.7138 @	0.018 +2	7.63 🛛	0.9949.		
ENRO +2	7.01 @	0.027 @	8.51 @	0.8858 @	0.015 @	7.37*	0.9902 «		
ORB +?	8.76	0.028	4.17*	0.9599 +	0.013 +	9.19	0.9918		

Validation parameters of the proposed method in a pork sample.										
Analytes₽	Linear range. Calibration equation.		Correlation coefficient	LOD	LOQs 🗸					
	(μg kg ⁻¹) _"		$(\mathbf{R}^2)_{\epsilon^2}$	(µg kg ⁻¹)₽	(µg kg ⁻¹)₽					
ENO _€ ²	2-200	$y = 2960x-1.38 \times 10^{3_{\phi^2}}$	0.9993+2	1.0+2	3.040					
OFL ⁴³	2-200	$y = 759x-1.03 \times 10^{3}$	0.9990+	0.5+2	1.70					
CIP	2-200	$y=1560x{+}1.21\times10^{4_{\phi}}$	<mark>0.9989</mark> ₽	0.8+2	2.6					
DAN.	2-200	$y = 1990x-1.21 \times 10^{3}$	<mark>0.9991</mark> ₽	0.7+2	2.3					
ENRO.	1-200+	$y = 1160x + 6.03 \times 10^{3_{e^2}}$	0.9995*	0.3+2	1.0					
ORB₄J	1-200+2	$y = 485x + 2.37 \times 10^{3}$	<mark>0.9999</mark> ₽	0.1+2	0.34					

Comparison of the pseudo-first-order and pseudo-second-order models for FQs.+



The typical chromatograms of (a) standard solutions with the concentrations of 100 μ g L⁻¹; (b) spiked (50 μ g kg⁻¹) pork sample after MSPE; (c) blank pork sample.

Analytes ₽	Added	Pork⊷			Chicken 42				Bovine 43				
	(µg/kg)₊ ^j	Found +	Recovery	MEª	RSD	Found	Recovery	MEª	RSD	Found₊	Recovery	MEª	RSD
		(µg kg ^{−1})+ ³	(%)₽	(%)₽	(%)₽	(µg kg ^{−1})⇔	(%)₽	(%)₽	(%)₽	(µg kg ^{−1})¢	(%)₽	(%)₽	(%)⇔
ENO 🕫	0₊⊃	ND43	сь С	¢,	C.	ND43	¢	C.	сь С	ND43	ę	ę	¢
	5₽	4.6+2	92.0₽	90.1∻	3.4∻	4.2∻	84.0₊3	88.8₽	3.2₽	4.3₽	86.0₽	95.7₽	4.1₽
	50≁	47.4₽	94.7₽	92.5₽	2.7₽	41.4∻	82.7∻	89.1₽	3.8₽	46.9₽	93.8₽	93.5₽	3.4₽
	100+7	95.6₽	95.6₽	94.5₽	4.6₽	ب 3.09	90.3 ₽	89.9₽	1.7₽	82.9₽	82.9⇔	94.1₽	4.5₽
OFL +3	0.⇔⊃	ND40	¢,	ę	G₽.	ND43	÷	c,	42	ND₽	c.	÷	4 ²
	5₽	4.8↔	96.0₽	98.7₽	4.7∛	4.8₽	96.4 ₽	90.4₽	5.5₽	4.6₽	92.3₽	98.6₽	2.7₽
	50∻	51.2₽	102.4*	97.2∻	3.1₽	49.6∻	99.2₽	91.7₽	6.1₽	42.9₽	85.9₽	97.9₽	3.7₽
	100+7	110.2+2	110.2+2	99.0₽	4.5₽	101.3+2	101.3+2	93.4₽	4.6₽	98.8₽	98.8₽	99.4∛	4.9₽
CIP↔	0.47	ND43	с.	¢,	C.	ND+3	42	C.	ت.	ND43	C ₽	÷	47
	5₽	4.6↔	93.3₽	94.2∻	3.7₽	4.7₽	94.5₽	85.6₽	5.3₽	4.4₽	\$8.0₽	90.4 ↔	3.5₽
	50≁	48.7∻	97.4₽	93.7₽	3.8₽	48.1∻	96.2+2	88.1+2	2.3₽	49.9₽	99.8₽	89.9₽	3.1₽
	100₊ਾ	101.7₽	101.7₽	92.9₽	2.5₽	98.7∻	98.7₽	87.5₽	4.4₽	98.7₽	98.7₽	91.8₽	2.7₽
DAN 🖓	0.	ND43	¢₽	42	C.	ND 43	¢,	C∌	Ç.	ND43	C₀	¢,	÷
	5₽	4.7₽	94.0₽	95.1₽	3.2₽	4.3₽	\$6.0₽	88.2₽	5.4₽	4.8∢⊃	96.0₽	91.6₽	3.4₽
	50≁	46.7₽	93.4₽	96.4₽	4.0∢	47.5₽	95.0₽	87.3₽	4.5₽	45.7₽	91.4.	94.2₽	5.7₽
	100*	96.5₽	96.5₽	93.8₽	5.1₽	⇔9.88	\$8.9₽	86.4⇔	2.7₽	93.2₽	93.2₽	93.9₽	5.5₽
ENRO.+3	0 ₊⊃	ND43	¢₽	¢,	ς	ND43	¢,	C.	⊊ _₽	ND43	C∌	сь.	¢
	5₽	4.6₽	92.0₽	99.2₽	5.7₽	4.2∻	84.0↔	92.5₽	3.9₽	4.8↔	96.0₽	97.2₽	4.0₽
	50≁	44.8₽	89.6₽	98.8₽	2.5₽	43.2₽	86.4₽	91.4₽	5.0₽	53.9₽	107.8+2	99.8₽	3.4₽
	100↩	91.5₽	91.5₽	97.1∻	3.2₽	87.5₽	87.5₽	94.3₽	4.2₽	82.2₽	82.2₽	97.4∻	4.5₽
ORB 🖓	0+3	ND↔	÷	¢,	C.	ND+3	÷	C.	47	ND43	C ₽	÷	4 ³
	5₽	4.5₽	90.0₽	89.3⇔	5.0₽	4.4₽	\$8.0 <i>+</i> ²	86.3₽	3.2₽	4.1₽	\$2.0₽	92.8↩	5.8₽
	50≁	44.4₽	\$8.8₽	90.9₽	3.8₽	45.5₽	91.0₊	85.9₽	3.7₽	47.3₽	94.6₽	91.9≁	6.0₽
	100≁	90.1∻	90.1+2	89.7₽	4.5₽	95.7₽	95.7₽	87.1₽	5.0₽	91.2₽	91.2₽	93.0₽	5.7₽

542 FQs analytical results in meat samples.

543 a Matrix effect

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TrAC Trends in Analytical Chemistry

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Recent advances and applications of polydopamine-derived adsorbents for sample pretreatment



IF= 8.428; TOP



Solid phase microextraction

IF= 8.428; TOP



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Recent advances in emerging nanomaterials based food sample pretreatment methods for food safety screening

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Conclusion

The sensitive determination of contaminants with low concentration in complicated foodstuffs is still a challenging work.

COFs with unique advantages have been considered as potential adsorbents in food sample pretreatment. But COFs still have many deficiencies as adsorbents and new COFs with functional structure, excellent adsorption ability and better selectivity to target compounds with different polarities should be further developed.

The combination of COFs with other functional materials may further enhance their adsorption ability

Recent Publications

First Author or Corresponding Author :

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- **2. Biosensors and Bioelectronics**, 2019, 111691. (**IF 9.518**)
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- 4. TrAC Trends in Analytical Chemistry, 2019, 115668. (IF 8.428)
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- **9.** Food Chemistry, 2019, 125455. (IF5.399)
- **10.** Biosensors and Bioelectronics, 2016, 79: 728-735. (IF 9.518)
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- **12. Biosensors and Bioelectronics** 85 (2016): 358-362. (IF 9.518)
- **13.** Analytica Chimica Acta, 2017,973: 91-99 (IF 5.256 top)
- **14. Food Chemistry,** 2017, 234(1): 408-415 (**IF 5.399 top**)
- 15. Journal of agricultural and food chemistry, 2018 (66): 3572-80. IF 3.41 ESI Hot paper
- **16. Microchimica Acta**, 2017, 184(7): 1923–1931 (**IF 5.479**)
- **17. Talanta**, 2017, <u>165</u>(1):677–684 (**IF 4.244**)
- 18. Talanta, 2018,190, 0039-9140 (IF 4.244)

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Thank you for your attentions !





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